



Research article

Nitrogen level affects growth and reactive oxygen scavenging of fenugreek irrigated with wastewater

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Abstract: Today, due to the constraint in availability of the freshwater for irrigation, wastewater is being used for irrigation of agriculture fields. Wastewater contain plant nutrients that favour crop growth but leave a burden of heavy metals which can enter the food chain and is a cause of great concern. This research work aimed at evaluating the potential utilization of wastewater as an alternative source of water and nitrogen (N) for fenugreek. For this purpose, a pot experiment was conducted with wastewater and four doses of nitrogen (0, 20, 40 and 60 kg.ha⁻¹) to observe a comparative effect on growth and biochemical characteristics using fenugreek as the test crop. Wastewater and soil used in the experiment were analyzed for their physicochemical properties. Results indicate that wastewater was richer in essential plant nutrients but contained some heavy metals in amounts well below the permissible limits for its use as irrigation water. Use of wastewater with 20 kg.N.ha⁻¹ improved growth, proline and antioxidant enzymes of the plant as compared to application of groundwater with no fertilizer. Lipid peroxidation increased with wastewater but decreased with the increasing nitrogen doses, so the best combination (WW×N₂₀) has negative impact on this parameter. These results concluded that wastewater can efficiently substitute groundwater and also works as a nutrient source whereas application of nitrogen could provide protection against the oxidative stress by increasing the antioxidant protective system.

Keywords: *Trigonella foenum-graecum* - Proline - Lipid peroxidation - Catalase - Peroxidase - Superoxide dismutase.

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INTRODUCTION

In many parts of the world freshwater resources is scarce, and there is with gradual destruction and increased pollution of fresh water resources, requiring alternatives to supply water for agriculture need to be considered (Yoshida & Ishii 1993). One solution to insufficient water supplies is water reuse. Sewage water for irrigation is a way to dispose of sewage; provide the reliability of large volumes of water for irrigation; to reduce discharge into potable water bodies; serve as an alternative source to chemical fertilizers, provide organic matter; improve soil infiltration capacity, permeability, cation and anion exchange capacities, water holding capacity and texture (Schneider & Erickson 1972, FAO 1992, Avemelech 1993, Brady & Weil 1999, Van der Hoek *et al.* 2002). However, there is a concern about contamination and bioaccumulation in plants of potentially toxic elements such as chromium (Cr), lead (Pb), nickel (Ni), cadmium (Cd), copper (Cu) and zinc (Zn) from domestic and industrial sources (Chen & Chia 2002, Kiziloglu *et al.* 2007).

In urban areas, especially in developing countries, raw wastewater for irrigation is used to grow vegetables for nearby urban markets (Lucho-Constantino *et al.* 2005). If the wastewater is used after proper dilution for irrigation hazards of heavy metals can be minimized (Pathak *et al.* 1999). The water used in the proposed research work has already been diluted from the source (Tak *et al.* 2012, Akhtar *et al.* 2012, Iqbal *et al.* 2012, 2017, Chalkoo *et al.* 2014, Sahay *et al.* 2015) and the contents of all the heavy metals except Ni were well within the permissible limits of Awasthi (2000) (Table 1). The presence of pathogenic microorganisms in wastewater poses an additional health-hazard. In order to avoid this problem, side-on-soil-surface watering was

used with watering can and practiced in pots to avoid direct contact with crop foliage. However, two weeks before the harvest, this technique was replaced by direct watering on the foliage (Cisse 1997, Akponikpe *et al.* 2011). Wastewater provides variable nutrients and improves attributes of the plant compared to fresh water, but the nutrient concentrations are not sufficient to fulfill requirements of the crop. External inputs need should be guided by the efficiency of their use, as the use of fertilizers beyond limits, in addition to being uneconomical, is potentially harmful to the environment by leaching through the soil beyond the root zone, eventually reaching ground water, or through surface runoff be deposited into nearby water bodies and cause eutrophication.

Table 1. Physicochemical characteristics of ground water (GW) and 100% wastewater (100% WW). All determinations in mg.l⁻¹ or as specified.

Determinations	GW	WW	FAO acceptable level
pH	7.30	8.00	6.5–8.4 ^a
EC (μ mhos cm ⁻¹)	700.00	1310.00	0.25–3.0 ^a
TS	955.00	1550.00	2200 ^c
TDS	560.00	1060.00	<2000 ^a
TSS	400.00	576.00	-
BOD	16.25	105.55	<25 ^b
COD	65.45	150.25	30–160 ^b
NO ₃ -N	0.83	2.75	<10.0 ^a
NH ₄ -N	0.11	4.45	5.0 ^a
Na ⁺	20.56	59.35	-
K ⁺	4.94	20.44	<2.0 ^a
Ca ⁺⁺	15.45	52.35	<400 ^a
Mg ⁺⁺	27.35	118.72	<61 ^a
Cl ⁻	55.25	113.22	<350 ^a
CO ₃ ⁻	54.54	147.74	-
HCO ₃ ⁻	105.33	80.48	<610 ^a
SO ₄ ⁻	30.86	88.95	-
PO ₄ ⁻⁻⁻	0.36	1.24	<2.0 ^a
Cu	-	0.209	0.20 ^d
Cd	-	0.009	0.01 ^d
Ni	-	0.331	0.20 ^d
Zn	-	0.153	2.0 ^d
Cr	-	0.039	0.10 ^d
Pb	-	0.028	5.0 ^d

Note: ^aAyers and Westcot, (1994); ^bPescod, (1992); ^cIndian Standard Institution (1974); ^dAwasthi, (2000).

In India, vegetable consumption is many times more than the European countries due to the vegetarian food habit of a large proportion of the population, and the lower cost of the common vegetables compared to meat, fish, egg and milk products. It is also important that, compared to other crops, many vegetables can be grown throughout the year and be marketed regularly, due to a short life-cycle.

Fenugreek (*Trigonella foenum-graecum* L.) is used fresh or dried, as spice (seeds) and vegetables (fresh leaves, sprouts and microgreens). Fenugreek leaves contain calcium, phosphorus and iron. Seeds are bitter in taste and have been known over 2500 years for their medicinal qualities (chronic dysentery, diarrhoea, chronic cough, enlargement of liver and spleen). The seeds contain the steroidal substance “diosgenin” which is used as a starting material in the synthesis of sex hormone as oral contraceptive. The study examined application of urban wastewater for irrigation and its effect on Fenugreek supplemented with nitrogen.

MATERIALS AND METHODS

Wastewater (WW) was collected in 50 L Jerry cans for watering the pots, as and when required, from the drain running along the Aligarh Mathura road, 6 km away from the Department. The water analyzed at the start and end of each experiment. Tap water was used as the control (GW) pumped directly from under the earth's surface and stored in syntax water tanks. The water types were analyzed for various physicochemical

characteristics (Table 1) following standard procedures (APHA 1998). Determination of heavy metals in the wastewater samples was with an Atomic Absorption Spectrophotometer (SENSAA GBC Avanta var. 2.02, Aligarh, India) according to the method of Ademoroti (1996). Microbiological examination of wastewater was conducted in the Microbiology Laboratory, Department of Agricultural Microbiology, Aligarh Muslim University, Aligarh, following quality guidelines of WHO (1989) and FAO (1992).

Before sowing, samples of the sandy loam soil were randomly accumulated from each pot, air-dried, mixed well, ground in a mortar and pestle, and passed through a 2 mm sieve before analysis. Soil pH and electrical conductivity were determined following the procedure of Jackson (1958). Organic matter was measured using the method described by Walkley & Black (1934). Estimation of carbonates and bicarbonates was followed Richards (1954). Nitrate nitrogen was estimated according to Ghosh *et al.* (1983). Phosphorus was measured following Olsen *et al.* (1954); potassium was estimated following Jackson (1958). Soil cation exchange capacity was determined following Ganguly (1951). Soil heavy metals were determined following Lindsey & Norwell (1978).

The experiment was conducted in the November 2010 to March 2011. The experiment was a completely randomized design replicated 3 times with main treatment of 100% urban wastewater (WW) and groundwater (GW). Doses of NPK fertilizer ($N_0P_0K_0$, $N_{20}P_{30}K_{30}$, $N_{40}P_{30}K_{30}$ and $N_{60}P_{30}K_{30}$) were mixed into the soil with the fertilizers from urea, single super phosphate and muriate of potash.

Effects of treatment were determined by measurement of plant: length, fresh and dry weights, numbers of leaves, leaf area, leaf fresh and dry weights, proline content, malondialdehyde content and enzymatic antioxidants (catalase, peroxidase and superoxide dismutase) at vegetative, flowering and fruiting stages at 30, 60 and 90 DAS, respectively. Plants were uprooted and lengths of the main tap root and shoot were measured. To assess dry mass, plants were dried for about 72 hrs in a hot air oven (forced air oven) maintained at 80°C and then weighed. After separating all the leaves at the petiole and stem junction, leaf area was measured using a leaf area meter (LA 211, Systronics, India). Proline content in fresh leaves was estimated following the procedure of Bates *et al.* (1973). The level of lipid peroxidation products in leaves was determined as malondialdehyde content by a modified version of the method described by Cakmak & Horst (1991). Catalase activity (EC 1.11.1.6) was measured according to Aebi (1984). SOD activity (EC 1.15.1.1) was estimated by recording the decrease in absorbance of superoxide nitroblue tetrazolium complex by the enzyme (Sen Gupta *et al.* 1993). The method of Bergmeyer *et al.* (1974) was used to measure oxidation of pyrogallol to purpurogallin by peroxidase.

The data were analyzed according to Panse & Sukhatme (1985) using a two-way analysis of variance after Steel & Torrie (1962) SPSS (ver. 11.0, Chicago, IL). If interactions were significant they were used to explain results using Duncan Multiple Range Test (Duncan 1955).

RESULTS

The physicochemical and microbiological properties of water types varied (Table 1). Electrical conductivity, total solids, total dissolved solids, BOD, COD, nitrate nitrogen, ammonia nitrogen, sodium, calcium, magnesium, chlorine, carbonate, sulphate and phosphate had higher values in wastewater than in groundwater (Table 1). Contents of Cu, Cd, Zn, Cr and Pb were within permissible limits; the Ni content exceeded permissible limits. The presence of microorganisms in wastewater from highest to lowest was faecal coliforms (7.2×10^2 cfu/100 mL), total heterotrophic bacteria (2.83×10^6 cfu/100 mL), total coliforms (1.9×10^3 cfu/100 mL) and *Salmonella-Shigella* sp. (1.3×10^2 cfu/100 mL).

Soil characteristics prior to the study and after the study varied (Table 2). The pH of the pre-sowing soil was desirable in agricultural soil. In pots irrigated with wastewater, the pH of soil extract decreased from before sowing to after harvesting. The organic carbon of wastewater irrigation was higher than well water and soil irrigated with wastewater had higher organic carbon than pre-sowing soil. This indicates that wastewater irrigation helps improve fertility after crop harvest. Wastewater provided essential nutrients to the soil and the crop because wastewater irrigated soil had more nitrogen, phosphorus and potassium than pre-sowing soil. Concentrations of heavy metals were within permissible limits (Awasthi 2000).

The wastewater and nitrogen combination, WW \times N₂₀, was most effective as it produced the longest plants, higher plant fresh and dry weights, more leaves, higher leaf area, and higher leaf fresh and dry weight at the 3 growth stages (Figs. 1, 2 & 3). There was an increase of 65.57% at 30 DAS, 35.06% at 60 DAS and 35.03% at 90 DAS in plant length over the GW \times N₀. For plant fresh weight, interactions were significant only at fruiting

stage, where $WW \times N_{20}$ and $WW \times N_{40}$ had similar values with increases of 35.14 and 29.13% over $GW \times N_0$, respectively. The best combination increased 54.89, 52.13 and 51.81% in plant dry weight over $GW \times N_0$ at the 3 growth stages. Among interactions, $WW \times N_{20}$ increased leaf number by 35.16, 35.58 and 35.53% over $GW \times N_0$ at the successive stages. For leaf area, $WW \times N_{20}$, produced increases of 34.95, 35.29 and 35.37% over $GW \times N_0$ at the 3 growth stages, respectively. Interactions were significant at vegetative and flowering stages in leaf fresh weight. The $WW \times N_{20}$ had the best causing increases of 40.27 and 39.13% over $GW \times N_0$ at the vegetative and flowering stages, respectively. This optimum combination recorded an increase of 39.2, 35.95 and 37.06% in leaf dry weight over $GW \times N_0$ at the 3 growth stages, respectively. The combinations $WW \times N_{20}$ and $WW \times N_{40}$ were similar. The $WW \times N_0$ combination was similar with $GW \times N_{20}$, $GW \times N_{40}$ and $GW \times N_{60}$. The linear regression curves worked out for plant height with leaf number and leaf area with leaf fresh weight favour the present findings (Fig. 4).

Table 2. Physicochemical characteristics of soil collected before sowing. All determinations in mg.l^{-1} in 1:5 (soil-water extract) or as specified.

Determinations	Pre-sowing soil	After harvest (soil layer irrigated with wastewater)	Normal Range
Texture	Sandy Loam	Sandy Loam	-
CEC (meq 100 g^{-1} soil)	3.32	3.92	-
pH	7.74	6.85	-
Organic Carbon (%)	0.419	1.15	-
EC (μ mhos cm^2)	242	-	-
TDS	758	-	-
$\text{NO}_3^- \text{N}$ (g kg^{-1} soil)	0.352	0.283	-
Phosphorus (g kg^{-1} soil)	0.109	0.151	-
Potassium	8.2	11.6	-
Calcium	27.14	23.89	-
Magnesium	16.49	12.94	-
Chloride	29.47	-	-
Carbonate	19.34	-	-
Bicarbonate	92.43	-	-
Sodium	14.11	15.02	-
Sulphate	17.32	-	-
Cu	-	1.11	135–270
Cd	-	0.22	3–6
Ni	-	0.75	75–150
Zn	-	2.09	300–600
Cr	-	0.053	-
Pb	-	0.72	250–500

Proline content increased with increasing nitrogen dose as well as with wastewater; the $WW \times N_{60}$ interaction caused an increase of 24.09, 27.27 and 22.84% over $GW \times N_0$ at 3 stages, respectively. Wastewater treated plants had increased malondialdehyde content over GW irrigated plant. However, malondialdehyde content declined with increase in nitrogen. Among combinations, $WW \times N_0$ and $WW \times N_{20}$ and $WW \times N_{40}$ were best and similar (Fig. 5).

Wastewater irrigation increased the activity of catalase, peroxidase and superoxide dismutase. Nitrogen dose also positively influenced the enzyme activity with most activity under N_{60} . The $WW \times N_{60}$ combination produced increases of 25.19, 20.74 and 22.58% in catalase over $GW \times N_0$ at the 3 samplings, respectively. However, this combination increased peroxidase by 24.56% at the vegetative stage, and 20.04 and 18.91% in superoxide dismutase at vegetative and flowering stages over $GW \times N_0$, respectively. The interactions were not significant at flowering and fruiting stages for peroxidase but were significant at the flowering stage for superoxide dismutase (Fig. 6). The proline content and antioxidant enzymes increased with the combination of $WW \times N_{60}$. The proline content was positively correlated with CAT ($r^2=0.9479$), POD ($r^2=0.8991$) and SOD ($r^2=0.9056$) (Fig. 7).

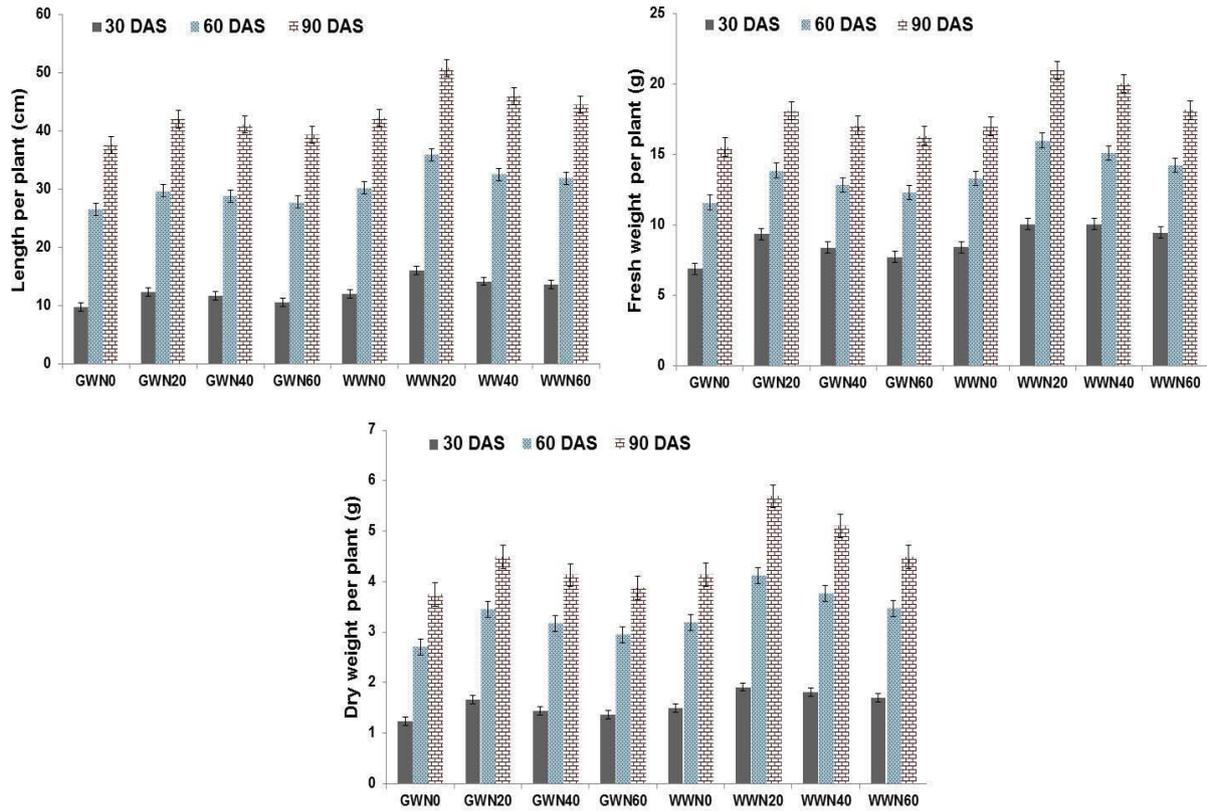


Figure 1. Effect of waste water on length, fresh weight and dry weight/plant of fenugreek supplemented with different nitrogen at 30, 60 and 90 DAS.

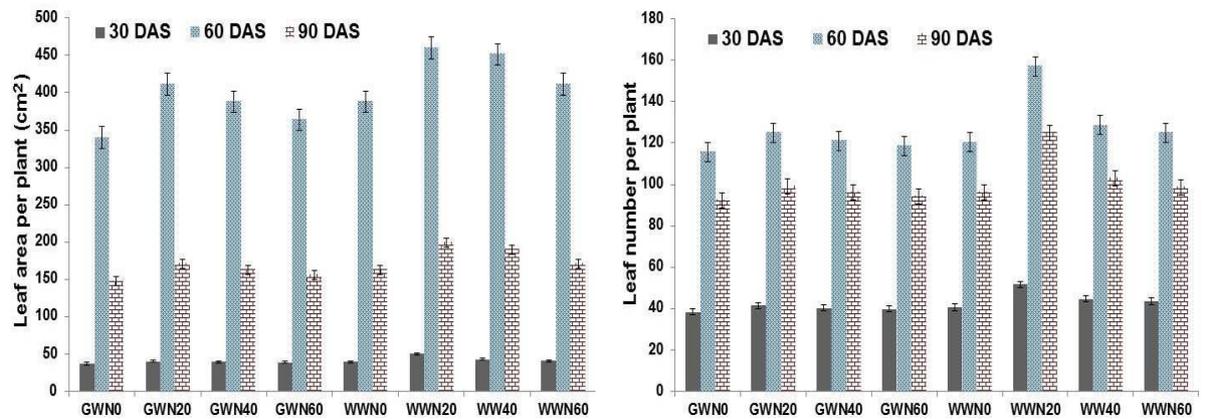


Figure 2. Effect of waste water on leaf number and leaf area/plant of fenugreek supplemented with nitrogen at 30, 60 and 90 DAS.

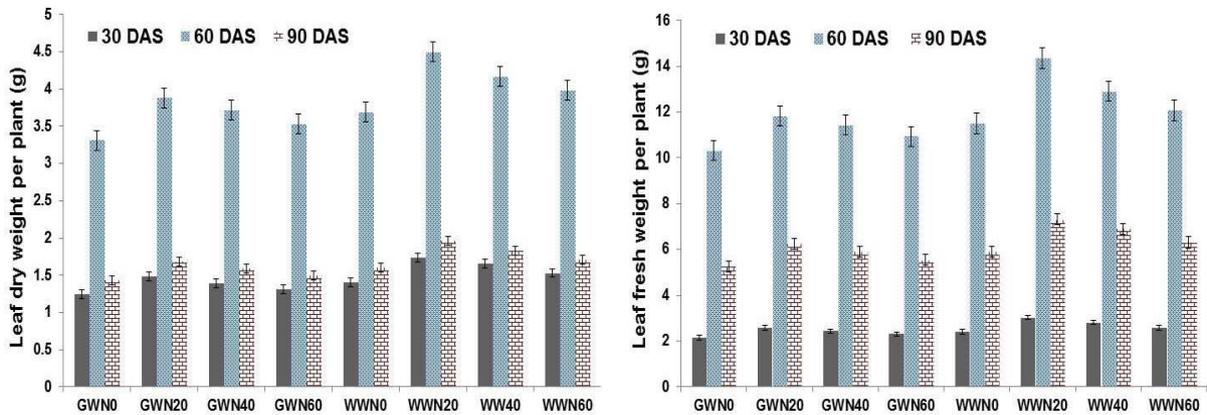


Figure 3. Effect of waste water on leaf fresh weight and leaf dry weight/plant of fenugreek supplemented with nitrogen levels at 30, 60 and 90 DAS (at 30 DAS, leaf dry weight×10).

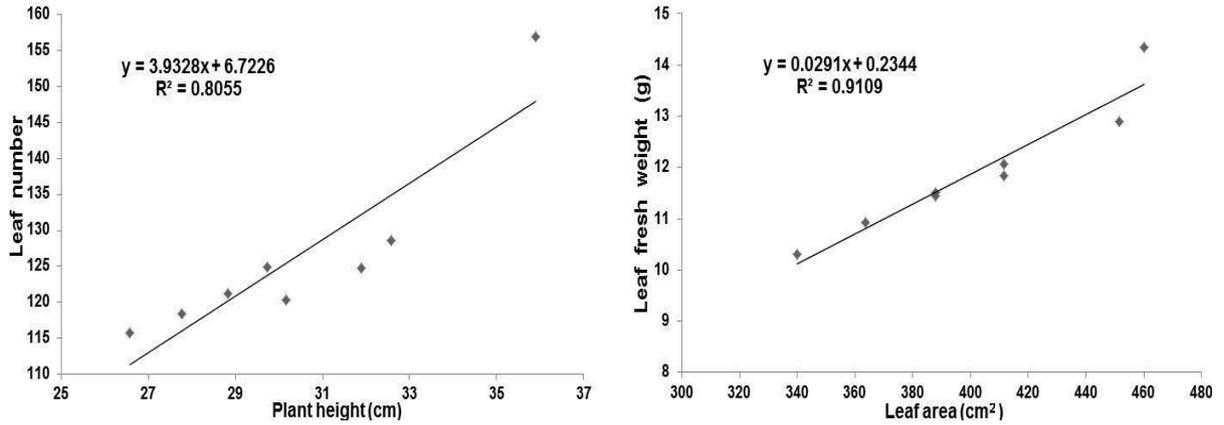


Figure 4. Linear regression curve of fenugreek showing correlation of plant height with leaf number and leaf area with leaf fresh weight at 60 DAS.

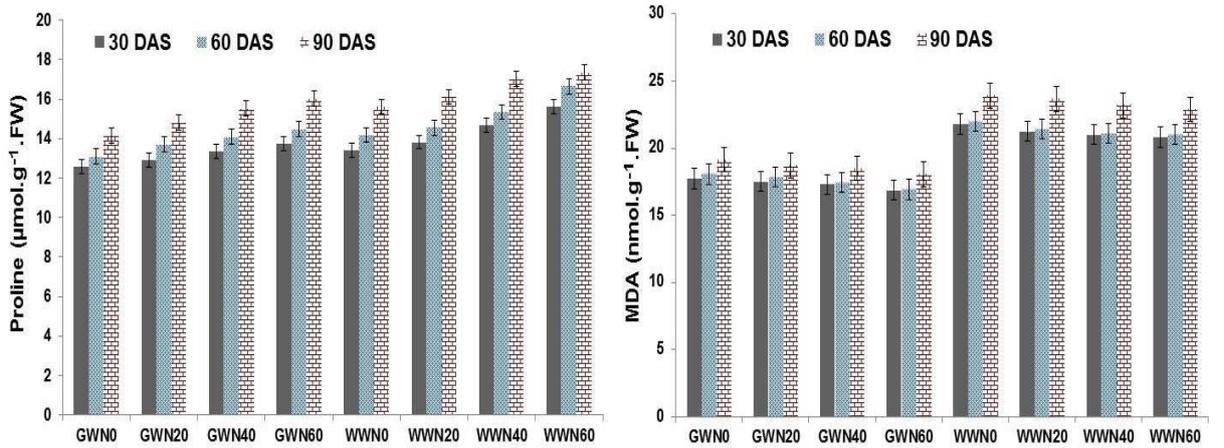


Figure 5. Effect of waste water on proline and malondialdehyde contents of fenugreek supplemented with nitrogen at 30, 60 and 90 DAS.

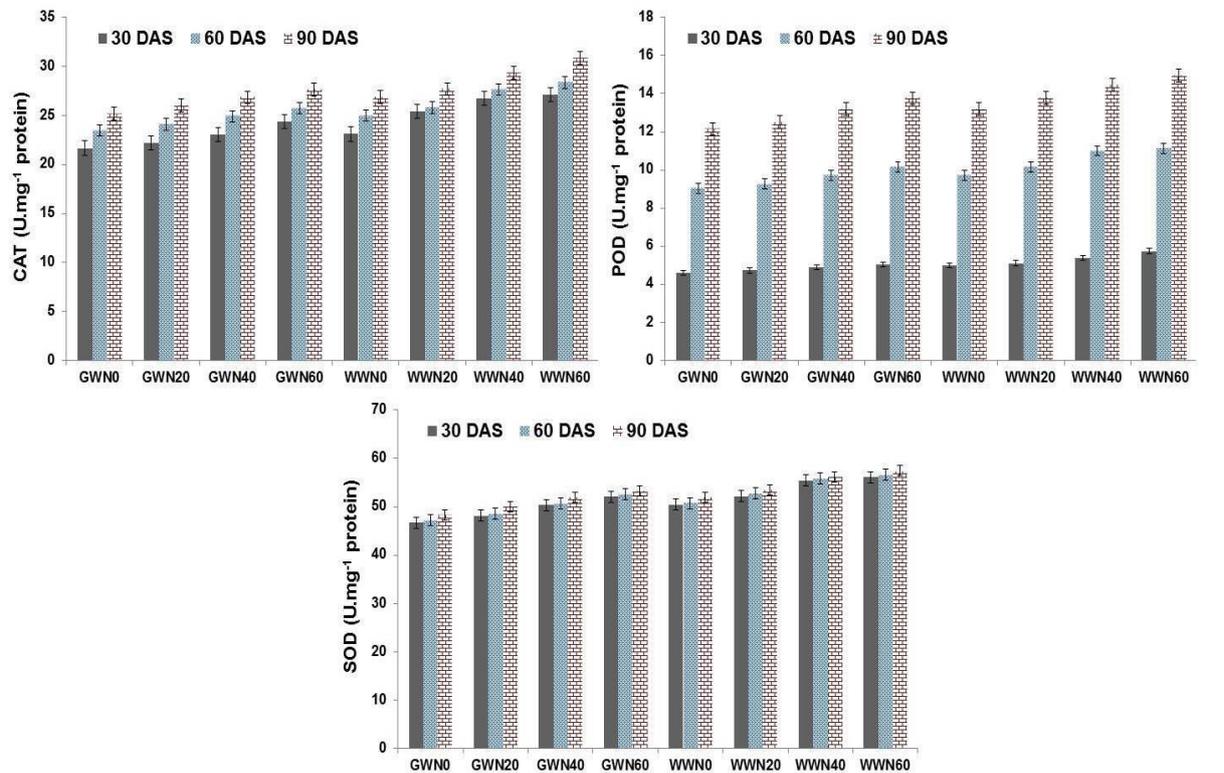


Figure 6. Effect of waste water on catalase, peroxidase and superoxide dismutase activity of fenugreek supplemented with nitrogen.

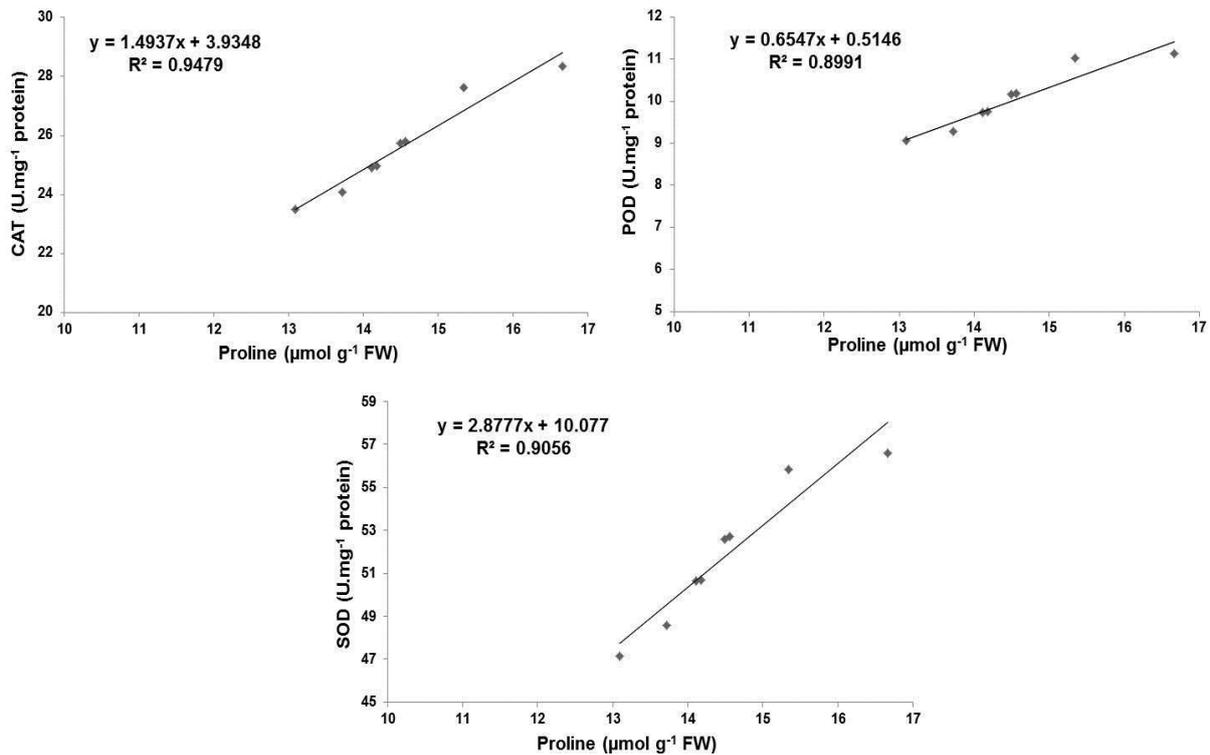


Figure 7. Linear regression curve of fenugreek showing correlation of proline with catalase, peroxidase and superoxide dismutase at 60 DAS.

DISCUSSION

The maximum values of various parameters were recorded with 100% concentration of wastewater which was black in color. The higher value of TDS is associated with higher EC of wastewater (Bhargava *et al.* 2008). The Cl^- , CO_3^{2-} and HCO_3^- were within the permissible limits (ISI 1974, Pescod 1992, Ayers & Westcot 1994). High BOD, COD, NO_3^- and PO_4^{3-} might be due to the presence of high oxidizable organic matter and rapid consumption of dissolved inorganic materials (Biswas *et al.* 2009, Kumar & Chopra 2012). The TSS content contributed to high salinity. Heavy metals at high levels in water bodies could cause toxicity to aquatic life (Zyadah & Abdul-Bakky 2000, Indra & Sivaji 2006) and buildup in river sediments (Singh *et al.* 1997). Total bacterial count and most probable number in wastewater are likely due to higher organic load of wastewater (Kumar & Chopra 2012). The pH is an important parameter as many nutrients are available for plant uptake only at a particular pH range. A pH value of 6.0–8.2 supports bacterial activity and is favourable for maximum yield. A shift in pH outside that range renders nutrients less available, even though they remain in the soil. Under acidic conditions, iron, aluminium, manganese, and heavy metals become highly soluble and may create problems for plants (Charman & Murphy 1991). The higher concentration of Na in the soil after effluent irrigation may be associated with the presence of high concentration of carbonate and bicarbonate in the effluent (Thompson *et al.* 2001). Reduction in water infiltration can occur when irrigation water contains high sodium relative to the calcium and magnesium content. This condition is called sodicity which reduces soil permeability (Kelley 1951). The overall increase in nitrogen is likely due to the nitrate and ammonium ions in wastewater. Irrigation with wastewater generally adds significant quantities of salts to the soil.

The beneficial effect on growth may be attributed to the nutrients present in wastewater (Table 1). However, reduced growth at higher concentrations may be due to the accumulation of salt which causes increased electrical conductivity (EC) with a link between EC and reduced plant growth (Kumar & Chopra 2012). The source of wastewater was diluted urban sewage which appears to be beneficial for the development of fenugreek because properly diluted wastewater can minimize the hazards of heavy metals (Pathak *et al.* 1999). Use of diluted wastewater from the same source produced beneficial effects on *Triticosecale* Wittm. (Shah *et al.* 2005), *Lens culinaris* Medikus (Tabassum *et al.* 2007a), *Brassica juncea* L. (Tabassum *et al.* 2007b), *Abelmoschus esculentus* (L.) Moench (Kausar 2009, Faizan *et al.* 2014, Kausar & Faizan 2015), *Cicer arietinum* L. (Tak *et al.* 2010, 2012, 2013), *Triticum aestivum* L. (Akhtar *et al.* 2012), *Capsicum annum* L. (Iqbal *et al.* 2012, Chalkoo *et al.* 2014, Iqbal *et al.* 2015).

Improvement in growth might be due to the presence of nitrogen. Fenugreek responded better to relatively low nitrogen dose because it is a leguminous crop and able to fulfill nitrogen needs by symbiotic nitrogen fixation. However, the process of symbiosis takes time to establish and became fully operative and the nitrogen accumulated in this manner is only available to the soil when the legume plant dies. During early plant growth, a low starter dose of nitrogen is necessary because higher doses hinder nodule formation (Mann 1968, Arrese-Igor *et al.* 1997, Ram & Verma 2001, Shahroz 2009). Other reports concerning nitrogen application on the same crop differ (Chaudhary 1999a, b, Sharma 2000, Thapa & Maity 2003, Datta *et al.* 2005, Tuncturk *et al.* 2011) where comparatively higher rates of nitrogen were reported.

Nitrogen is an important nutrient required in large amounts by plants. It is an integral constituent of proteins, nucleic acids, chlorophyll, co-enzymes, phytohormones and secondary metabolites. The availability of N to roots is therefore a decisive factor for plant growth (Marschner 2012). The urban wastewater used for irrigation contained nitrogen in the ionic forms, NH_4^+ and NO_3^- and further supplementation of nitrogen in the form of urea yielded nitrate rapidly within a few days which is absorbed by plants possibly contributing to a cumulative effect on growth (James 2010).

Proline is an amino acid that easily accumulates in plants under stress and is a marker of imbalance in a metabolic pathway. It also functions as an osmolyte to alleviate damage caused by biotic or abiotic stresses (Ashraf & Foolad 2007, Chandra *et al.* 2009, Myriam *et al.* 2009, Bedouh & Bekhouche 2012, Kumar *et al.* 2012a, b). Resistant plants accumulate more proline than less resistant plants (Naidu *et al.* 1992). Accumulation of proline preserves the structure and activity of protein, reduces enzyme denaturation and protects biomembranes from damage by inactivating hydroxyl radicals or other reactive chemical species (Pollard & Wyn Jones 1979, Smirnoff & Cumbes 1989, Saradhi *et al.* 1995). Proline accumulates in plants exposed to heavy metal stress (Zhao 2011, Baudh & Singh 2012a, b). Fenugreek irrigated with wastewater exhibited increase in the proline content. The wastewater contained heavy metals that cause stress to plants likely increasing proline content. Heavy metal induced proline accumulation has been reported (Lalk & Dorfling 1985, Bhattacharjee & Mukherjee 1994, De & Mukherjee 1998, Handique & Handique 2009, Kumar *et al.* 2015). Proline accumulation helps the plant to withstand the heavy metal stress by protecting the key enzyme such as glucose-6-phosphate dehydrogenase and nitrate reductase from being inactivated by toxic heavy metal ions by forming metal-proline complex (Handique & Handique 2009). In this study, proline content increased with increased nitrogen application. This may be because it is a nitrogen-storage compound (Ahmad & Hellebust 1988) and that synthesis and accumulation of proline are stimulated by nitrogen supply (Sanchez *et al.* 2002). This result is in agreement with the findings of many workers (Naidoo & Naidoo 2001, Sanchez *et al.* 2001, 2002, Zhao & Liu 2009, Ahmadi *et al.* 2010, Wang *et al.* 2011).

Malondialdehyde content (MDA) or thiobarbituric acid reactive substances (TBARS) is the decomposition product of polyunsaturated fatty acids of biomembranes obtained when reactive oxygen species (ROS) initiates the process of lipid peroxidation, which is considered an indication of oxidative stress in plants (Gosset *et al.* 1994, Blokhina *et al.* 2003). Production of the ROS superoxide radical (O_2^-), hydrogen peroxide (H_2O_2) and hydroxyl radical (OH^\cdot) increases as a result of abiotic stress such as can occur with heavy metal toxicity, salinity and drought in plants. Under normal conditions, these ROS are well regulated by cell metabolism but under abiotic stress, they may exceed the scavenging capacity of the antioxidant system. The wastewater contained heavy metals which may enhance the level of H_2O_2 which can result in the formation of hydroxyl radicals that can cause lipid peroxidation (Loggini *et al.* 1999). Oxidative damage to bio-membranes of plants can be primarily attributed to the presence of Cd, a redox active metal catalyzing the generation of hydroxyl radicals and O_2^- (Arora *et al.* 2002). The non-redox metals Zn, Ni, Cr, Cu and Pb do not produce ROS directly but generate oxidative stress by interfering with the plant's antioxidant defence system (Garczarska & Ratajczak 2000, Aravind & Prasad 2003, Panda & Choudhary 2005). These metals might have contributed to increased MDA levels (Olorunfemi & Lolodi 2011). Reduction of MDA content due to nitrogen application was probably due to increased activity of antioxidant enzymes (Shin *et al.* 2005, Zhang *et al.* 2007, Zhao & Liu 2009, Lin *et al.* 2011). Xiao *et al.* (1998) reported that nitrogen application can improve light reaction of photosynthesis which ultimately leads to the lower ROS production.

Biotic and abiotic stresses affect plants by causing excess accumulation of ROS, which can react with certain biomolecules, and alter or inactivate biochemical activities (Mittler 2002, Mittler *et al.* 2004, Choudhary *et al.* 2012). Even under optimal conditions, ROS are generated as by-products of normal metabolism in subcellular components (Asada & Takahashi 1987, Elstner 1991, Asada 1994, del Rio *et al.* 1998). These radicals are

capable of initiating peroxidation reaction (Bandyopadhyay *et al.* 1999). The ROS also influence degeneration of DNA, oxidation of deoxyribose sugar, the reforming and breaking of DNA strands and cause mutations (Zhang & Kirkham 1996, Chirkova *et al.* 1998). To migrate and repair damage initiated by ROS, plants developed complex antioxidant systems (del Rio *et al.* 2002). This defense system includes enzymatic and non-enzymatic components. The enzymatic defense system includes the enzymes superoxide dismutase (SOD), catalase (CAT), ascorbic peroxidase (APX), dehydroascorbate reductase (DHAR), and glutathione reductase (GR) (Yordanova *et al.* 2003). SOD is an enzyme acting as a first line of defence in combating oxidative stress in plants, which dismutates superoxide anions to H₂O₂ (Shalini & Dubey 2003, Mishra *et al.* 2006, Davies *et al.* 2009). CAT and POD are involved in building resistance against accumulation and toxicity of hydrogen peroxide by converting H₂O₂ to water and oxygen in cells (Msttes 2000). Increases in levels of all antioxidant enzymes occurred in fenugreek irrigated with wastewater. This is likely due to the pollutants present that can lead to oxidative stress causing biochemical changes (Lee *et al.* 2001, Panda & Upadhyay 2003, Singh *et al.* 2003, Fatima & Ahmad 2005, Baghel 2008, Xu *et al.* 2010, Olorunfemi & Lolodi 2011, Jain & Srivastava 2012, Sangeetha *et al.* 2012). Application of levels of nitrogen increased CAT, POD and SOD activities in fenugreek. This may be due to the alleviation of heavy metal damage by the nitrogen increased activity of antioxidant enzymes. Williams *et al.* (1967) suggested that addition of nitrogen in soil increases the formation of NO₃⁻ which decreases soil pH. Lowering pH increases the solubility of heavy metals and decrease the efficiency of adsorption (Moreno *et al.* 2000, Zhang *et al.* 2007, Zhao & Liu 2009, Ahmadi *et al.* 2010, Lin *et al.* 2011).

CONCLUSION

From the present study we concluded that the wastewater used for irrigation is suitable for fenugreek as the nutritive and toxic chemical elements present in wastewater were accumulated in the soil at low level and this accumulation did not cause any problem to the plants tested. Use of wastewater can provide economic benefits. Substitution of ground water by wastewater at some contents increased growth, proline, malondialdehyde content and antioxidant enzymes. Integrated application of wastewater and fertilizer increased growth and biochemical characteristics in fenugreek. Even if oxidative stress is induced in fenugreek plants irrigated with wastewater, application of N could provide protection against oxidative stress by increasing the antioxidant protective system.

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